Various topics Grand canonical Monte Carlo

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### Grand canonical Monte Carlo simulation

grand canonical ensemble:  $\mu$ VT (compare with canonical ensemble: NVT) constant chemical potential, variable number of particles

# Grand canonical Monte Carlo simulation

grand canonical ensemble:  $\mu VT$ 

(compare with canonical ensemble: NVT)

constant chemical potential, variable number of particles

GCMC

 explicitly accounts for density fluctuations at fixed volume and temperature

trial insertions and deletions of molecules

└─ Monte Carlo

# Grand canonical Monte Carlo simulation

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choose randomly if a particle insertion or deletion is attempted

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calculation of the acceptance probability:

$$\begin{split} \mathcal{P}(N \to N+1) &= \frac{V\Lambda^{-3}}{N+1} \cdot \exp[\beta\mu] \cdot \exp[-\beta(U_{N+1}-U_N)] \\ \mathcal{P}(N \to N-1) &= \frac{N}{V\Lambda^{-3}} \cdot \exp[-\beta\mu] \cdot \exp[-\beta(U_{N-1}-U_N)] \\ (\beta = \frac{1}{k_{\rm B}T}, \text{ de Broglie thermal wavelength } \Lambda = \sqrt{\frac{\hbar^2}{2\pi m k_{\rm B}T}}) \\ \text{note: practical implementations differ a little} \end{split}$$

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Various topics

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#### Grand canonical Monte Carlo simulation

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Applications:

└─ Monte Carlo

# Grand canonical Monte Carlo simulation

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■ interfaces – e.g. studies of adsorption

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# Grand canonical Monte Carlo simulation

Applications:

- interfaces e.g. studies of adsorption
- protonation states of amino acid side chains in a protein
  chemical potential of protons is related to pH

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### Grand canonical Monte Carlo simulation

Applications:

- interfaces e.g. studies of adsorption
- protonation states of amino acid side chains in a protein
  chemical potential of protons is related to pH

water molecules in a binding pocket / another cavity
work with the chemical potential of water