

Various topics

Coarse graining; hard bodies; Monte Carlo techniques

Marcus Elstner and Tomáš Kubař

June 15, 2016

United-atom force fields

- early biomolecular FF (e.g. Weiner84), popular in the 1990's
- hydrogen atoms considered as **condensed** to the heavy atom
- mass and charge represent such a group of atoms as a whole
- number of atoms reduced considerably relative to **all-atom** FF

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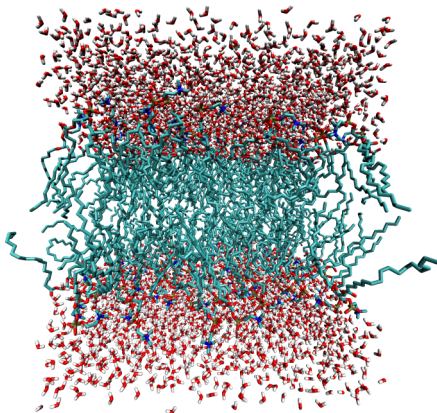
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- mass and charge represent such a group of atoms as a whole
- number of atoms reduced considerably relative to **all-atom** FF
- good for non-polar C–H bonds – so CH₃ is one united atom
- polar O–H group by a single 'atom' – too crude
 - only non-polar hydrogens usually condensed with heavy

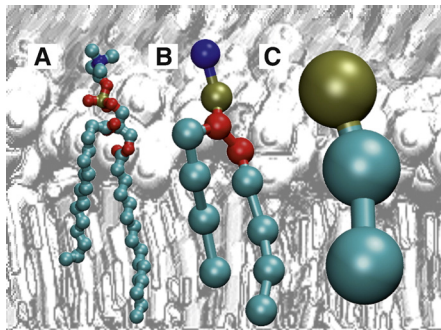
United-atom force fields

still sometimes used e.g. for lipids – each CH_2 is a united atom



(simulation of a DOPC bilayer in water – Berger FF for the lipid)

United-atom and coarse-grained force fields



(A) united-atom, (B) specific and (C) generic coarse-grained

from Marrink et al., Biochim. Biophys. Acta 2009

Coarse-grained models

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Considered are particles composed of **several** atoms – **beads**

Fewer inter-particle interactions → reduced computational expense

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The necessary parameters – often obtained by fitting to all-atom force fields

Coarse-grained models

Every bead usually represents several atoms,
and a molecule is composed of several beads

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– e.g. secondary structure of proteins is fixed with Martini FF

Also, hydrogen bonding cannot be described with beads!

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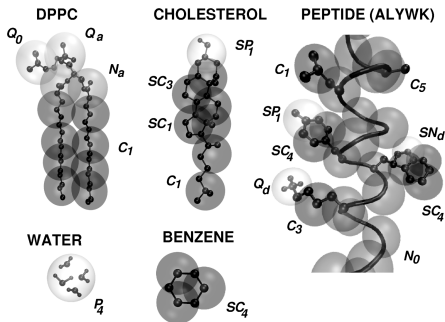
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solution – compensation with Lennard-Jones contributions

Especially useful for large-scale conformational transitions involving exceedingly large molecular systems, excessive time scales, or both

Martini force field

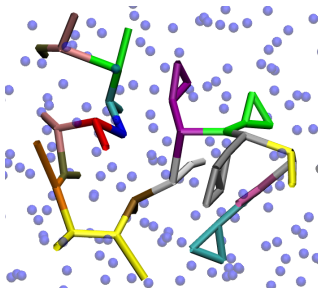
mapping of beads onto molecular fragments with Martini FF



- 3 to 4 heavy atoms compose one bead ('4-to-1 mapping')
- mass of beads – 72 u ($= 4 \text{ H}_2\text{O}$), or 45 u in ring structures

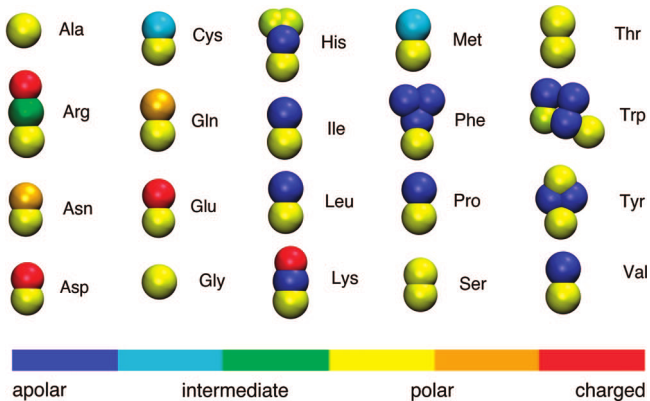
Martini force field

a solvated peptide with Martini FF



Martini force field

the amino acids:



Acceleration of the simulation

Why does a coarse-grained simulation run faster?

- smaller number of particles → fewer interactions to compute

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 - 25 fs with Martini (i.e. 100 fs effectively, see below)

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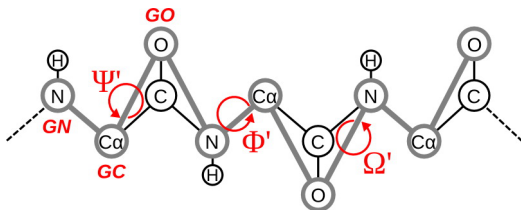
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“... length and time scales that are 2 to 3 orders of magnitude larger compared to atomistic simulations, providing a bridge between the atomistic and the mesoscopic scale.”

Coarse-grained models

SIRAH force field

- somewhat less coarse-grained, closer to united-atom
- representation of backbone dihedral angles retained















from Pantano et al., J. Chem. Theory. Comput. 2015

Coarse-grained models

SIRAH force field

- less coarse-grained → possibly improved transferability
- explicit solvent, long-range electrostatics (no cut-off)

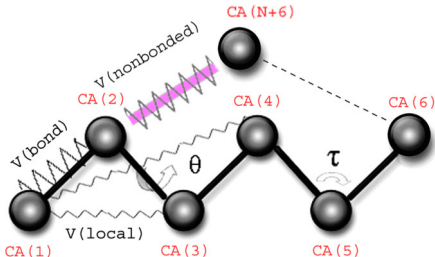
FG	CG	SIRAH name	q (e)	σ (nm)	ϵ (kJ/mol)	FG	CG	SIRAH name	q (e)	σ (nm)	ϵ (kJ/mol)
G 		1: GC	0,10	0,40	0,55	W 		4: BCG	0	0,35	1,70
		2: GN	0,125	0,40	0,55			5: BNE	-0,10	0,35	0,10
		3: GO	-0,225	0,40	0,55			6: BPE	0,10	0,35	0,01
S 		4: BOG	-0,20	0,41	0,35			E 		4: BCD	-0,30
		5: BPG	0,20	0,40	0,01	5: BOE1	-0,35			0,45	0,55
						6: BOE2	-0,35			0,45	0,55
Na ⁺ and 6 water molecules 		1: NaW	1,00	0,58	0,55	WT4 11 water molecules 				1: WN1	-0,41
				2: WN2	-0,41			0,42	0,55		
				3: WP1	0,41			0,42	0,55		
				4: WP2	0,41			0,42	0,55		

- illustration – different compromises may be made

Coarse-grained models

VAMM force field for proteins

- every amino acid represented by a single bead at C_α



- more coarse-grained than Martini

from Korkut & Hendrickson 2009

MD simulation of hard bodies

first MD simulation of a system in the condensed phase

- used the model of **hard spheres**
(Alder & Wainwright, J. Chem. Phys. 1957)
- first step from the ideal gas towards realistic molecules
- valuable tool in statistical thermodynamics
→ equations of state and virial expansions

The hard-sphere potential

- pairwise potential
- potential energy of a system of two hard spheres with radius R is zero for distances larger than the diameter of the spheres is infinity for shorter distances, when the spheres overlap:

$$V(r) = \begin{cases} 0 & \text{if } r > 2R \\ +\infty & \text{otherwise} \end{cases}$$

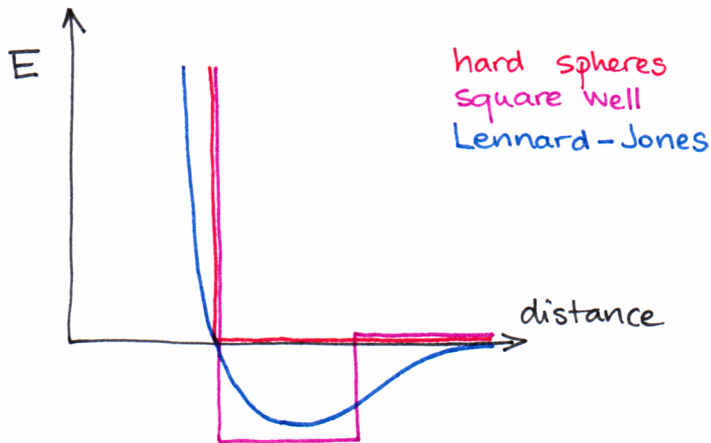
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- is **discontinuous** → not differentiable
- different from potentials typically used in biomolecular simulation

The hard-sphere potential



The square-well potential

a more realistic description preserving the simplicity of the model?

square well model

- region of negative potential energy (attractive interaction) starting at the contact distance $2R$
- goes in the direction of the Lennard-Jones potential, which describes nonpolar fluids very well

Hard convex bodies

- another extension used in statistical thermodynamics
- potential energy function is discontinuous, still:
zero if the bodies do not intersect; infinity if they do

Hard convex bodies

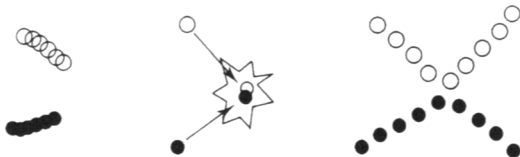
- another extension used in statistical thermodynamics
- potential energy function is discontinuous, still:
zero if the bodies do not intersect; infinity if they do
- enhancement – the bodies are not spherical anymore,
but rather ellipsoidal or polyhedral
- may describe e.g. diatomic molecules better than hard sphere

Simulation protocol

propagation of Newton's EOM with e.g. Verlet integrator
– continuous and smooth potential required
otherwise – sudden 'jumps' in forces lead to unstable simulations,
or at least wrong sampling of the configuration space

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reprinted from Leach, Molecular Modelling

Simulation protocol

Hard spheres cannot be simulated with a usual integrator

- explosions caused by sudden clashes of atoms would occur (similar to those in usual MD simulations with too large Δt)

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What would a simulation of hard spheres with Verlet look like?

There are no forces in any initial configuration,
and so the spheres move with their initial velocities
until, all of a sudden, two spheres start to overlap.

The energy and forces are infinite, and the simulation crashes.

Simulation protocol

The protocol has to be adjusted to the discontinuous potential

- event-driven protocol

The spheres move along straight lines between collisions,
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- 1 Identify the next pair of spheres to collide,
and calculate when this collision will occur
- 2 Calculate the positions of all spheres at the collision time –
conservation of linear momentum and of kinetic energy
- 3 Determine the new velocities of the two spheres after collision
- 4 Repeat from start

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No further approximations are involved in this protocol

→ simulation will be **exact** within the model of hard spheres

Note: With continuous potentials, we **had to** make approximations, like a stepwise integration of the eqns of motion

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Potential energy – constant (zero) throughout the simulation

Conservation of total energy → conservation of kinetic energy

→ temperature is constant in any hard-spheres simulation

Monte Carlo simulation

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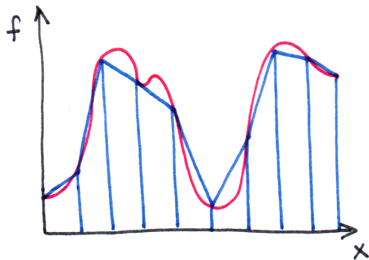
Another possibility – Monte Carlo methods (MC),
which involve random number generators Actually, first
computer simulations of molecular systems were **MC**
(Metropolis et al., J. Chem. Phys. 1953)

Monte Carlo integration

Major goal of molecular simulation – calculation
of thermodynamic properties – integration (formally)
Can we use a method based on randomness for integration?

Monte Carlo integration

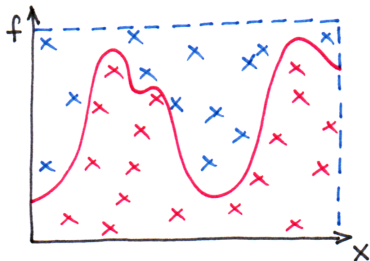
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- Possibility – trapezium rule
- comes into trouble
for functions of many variables
 - we always have many variables
molecular systems

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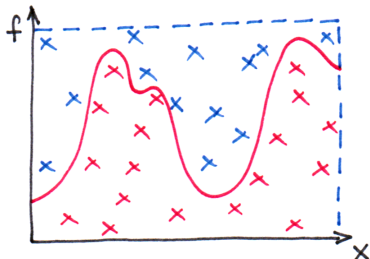


Alternatively

- generate N points randomly
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- area under the curve relative to the rectangle $\approx n/N$

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Apply the Monte Carlo idea to calculate π as follows:

Generate pairs of random number between 0 and 1 (x, y).

Count the pairs for which $x^2 + y^2 < 1$, i.e. the point (x, y) lies within the circle centered at (0,0) with a radius of 1.

The ratio of this number to the total number of pairs approaches $\pi/4$.

Monte Carlo integration

Importantly:

Extension of this ansatz to many dimensions is straightforward

- useful for studies of molecular systems

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Groundbreaking idea (Metropolis):

Generate the configurations with the **right probability**,
creating the correct thermodynamic (e.g. canonical) **ensemble**

Such **importance sampling** will make it trivial to average
thermodynamics quantities over the generated configurations

Metropolis' method

Typical MC simulation of a molecular system

- a sequence of configurations is generated in an iterative way
- in every iteration, one configuration is produced.

Usually:

- 1 A **trial** configuration is constructed from the current one by randomly shifting one randomly chosen particle (atom).

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- 1 A **trial** configuration is constructed from the current one by randomly shifting one randomly chosen particle (atom).
- 2 It is tested if this configuration shall be **accepted or not**.
For this, potential energy of the entire system is calculated.
(possible optimization – only small part of the system changes,
→ only a small fraction of the interactions changes)

Metropolis' method

Trial set of coordinates is calculated with random $\xi_{x,z,y} \in (0, 1)$:

$$x_{\text{trial}} = x + (2\xi_x - 1) \cdot \delta r$$

$$y_{\text{trial}} = y + (2\xi_y - 1) \cdot \delta r$$

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The trial configuration is **accepted** if $\mathcal{P} > \text{random } \zeta \in (0, 1)$ otherwise it is **discarded** and another trial is generated

Acceptance ratio

The percentage of accepted configurations (among all generated) governed by max. allowed displacement δr – adjustable parameter

- usually chosen so that $\frac{1}{3}$ to $\frac{1}{2}$ of all configs are accepted
- this was shown to lead to the most efficient sampling

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δr too small \rightarrow most configurations are accepted though,
but the configurations are very similar \rightarrow slow sampling

δr too large \rightarrow too many trial configurations are rejected

Often – δr adjusted in the course of the simulation

in order to reach a certain target acceptance ratio

Properties of MC

- generates a correct thermodynamic ensemble (canonical)
- involves **temperature** naturally
 - no additional thermostat necessary
 - difference from MD
- no kinetic information (velocities, E_{kin})

MC protocol – variations

Possible modifications to the algorithm:

- move the atoms sequentially, in a preset order, instead of selecting one randomly – one fewer random number needed
- move several atoms at once, instead of a single atom
 - very efficient sampling of config space (with appropriate δr)

Generators of pseudorandom numbers

Several random numbers in every iteration have to be obtained
and a large number of iterations is needed
→ reliable and efficient source of random numbers needed.

Most convenient – ‘calculate’ random numbers in some way
paradoxical requirement (computers are deterministic)

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There are ways to generate sequences of **pseudorandom numbers**
not actually random, but still independent enough of each other,
with right statistical properties → useful for MC

Linear congruential generators

- most commonly used generators
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value $\in (0, 1)$ is obtained by dividing ξ_{i+1} by the modulus m

Linear congruential generators

Very important – choose appropriate values of a , b and m

Then, the generator will produce all possible values $0, \dots, m - 1$
and will start to repeat the sequence only after m numbers.

Otherwise – the sequence starts to repeat itself much earlier,
and the randomness is severely limited.

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Disadvantage – if we generate points in an N -dimensional space,
these are not distributed uniformly in the space,
but rather they lie on at most $\sqrt[N]{m}$ $(N - 1)$ -dimensional planes
(i.e. on straight lines if we have a 2D space).

With really poor generators – much fewer than $\sqrt[N]{m}$ hyperplanes.

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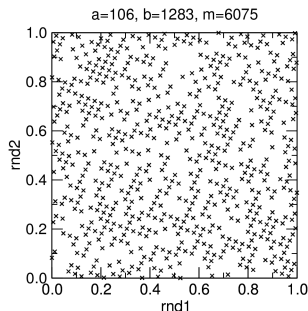
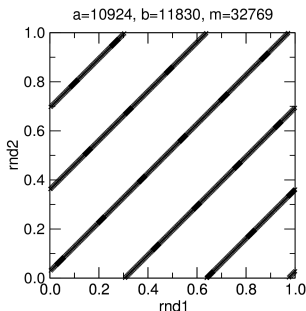
An example is RANDU: ξ_0 is odd and $\xi_{i+1} = 65539 \cdot \xi_i \bmod 2^{31}$.

All generated values are odd, the period is only 2^{29} ,

and the points $(\xi_i, \xi_{i+1}, \xi_{i+2})$ cumulate on as few as 15 planes in space.

Linear congruential generators

A bad and a good generator of pseudorandom numbers.



Each point $(rnd1, rnd2)$ is a pair of consecutive numbers from LCG

Generators of higher quality

Still, LCG are often used in MC simulations
because of extreme simplicity and computational efficiency.

Higher-quality pseudorandom number generators:

linear feedback shift register generators

- uses several bits from current number to generate new ones
- does not cumulate the generated numbers on hyperplanes

Mersenne twister

- current state of the art among generators
- extremely long period of $2^{19937} - 1$
- no cumulation of numbers on hyperplanes up to 623 dim.
- even suitable for cryptographic applications

Alternative generators of random numbers

In Unix-like operating systems (with Linux being the first),
 /dev/random (or /dev/urandom) is a special file
 that serves as a random or pseudorandom number generator.
It accesses environmental noise collected from device drivers etc.

from Wikipedia

Monte Carlo simulation of molecules

Easiest implementation – system of monoatomic molecules
(translational degrees of freedom only)

Polyatomic molecules – more complex situation,
most difficult if there is large conformational flexibility

Monte Carlo simulation of molecules

Easiest implementation – system of monoatomic molecules
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→ overlap of atoms → energy grows steeply

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Rigid molecules – still quite easy to simulate with MC

– orientation in space being varied beside position in space

– rotation along an axis x , y or z by randomly chosen angle

Monte Carlo simulation of molecules

Particularly rich area of application of MC
– macromolecular chemistry

Monte Carlo simulation of polymers

Many approximative polymer models are suitable for MC simulation
Convenient – a chain of monomer units,
which are elementary particles (no further internal structure)

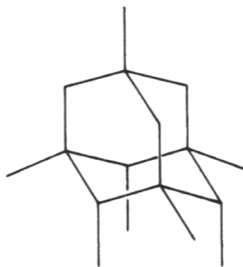
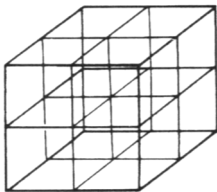
Monte Carlo simulation of polymers

Many approximative polymer models are suitable for MC simulation

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Lattice models – very simple and useful for efficient studies

- monomer units connected with a bond can occupy neighboring lattice points in a cubic or tetrahedral lattice

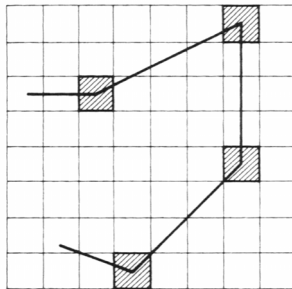
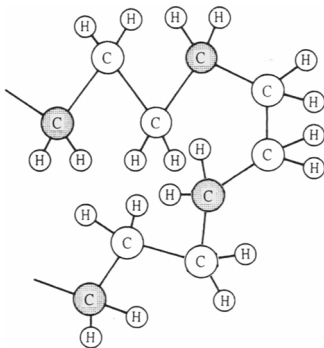


Monte Carlo simulation of polymers

Usually – very simple expressions for potential energy
(simplicity of the model + requirement of efficiency)

More realistic and complex – **bond fluctuation** model

- lattice is finer-grained compared to the bond length
- ‘effective’ bonds are not constrained to the edges of lattice



Monte Carlo simulation of polymers

Simplest type of simulation – **random walk**

- the chain grows in a random direction until the desired length
- first implementation – excluded volume of previous segments is not considered, so the chain is free to cross itself

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Various structural properties can be evaluated

by averaging the results over many growing ‘simulations’.

Quantities of interest: end-to-end distance R_n , radius of gyration s_n

$$\begin{aligned}\langle R_n^2 \rangle_0 &= n \cdot l^2 \\ \langle s^2 \rangle_0 &= \langle R_n^2 \rangle / 6\end{aligned}$$

for a chain composed of n bonds with length l

Monte Carlo simulation of polymers

Excluded volume not described – may seem to be crude,
but this may not be a problem

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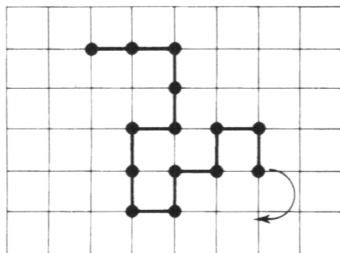
theta state (ϑ state) of a polymer

- excluded volume and attractive interactions cancel exactly
(also, the second virial coefficient vanishes)
- results derived with simple random walk are actually valid
(often designated with the subscript '0')

Monte Carlo simulation of polymers

How to take the excluded volume into account?

- do not allowing the chain to extend to already occupied points
- self-avoiding walk



Monte Carlo simulation of polymers

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SAW was used to generate all possible configurations of a polymer of given length on a given lattice

→ partition function → all thermodynamic properties

‘potential energy’ – simple interaction model for nearby monomers
also – copolymers with two different types of monomer units

particular attention – structural properties – end-to-end distance:

$$\langle R_n^2 \rangle \approx n^{1.18} \cdot l^2 \quad \text{for } n \rightarrow \infty$$

Monte Carlo simulation of polymers

to modify the configuration with MC – nearly impossible
because no free space (espec. for densely packed polymers)

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slithering snake model – a trick to avoid the problem:

1. choose one end of the polymer chain randomly as the head
2. make MC attempt to connect a new monomer unit to head
3. on success, then remove one monomer from the other end
4. repeat

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How to improve the lattice models? ... cancel the lattice

- ‘**continuous**’ **polymer models** – string of connected **beads**
- freely connected, with a spherically symmetric potential (LJ)
- do not generally correspond to monomer units
- links (‘bonds’) either fixed or with a harmonic potential

Monte Carlo simulation of polymers

most unrealistic property – continuous variation of link angles

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- free rotation about the links ('dihedral angles') allowed
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- free rotation about the links ('dihedral angles') allowed
- this changes the overall structure of the polymer chain!
- the **extension** of a chain with interlink angle φ ,
with respect to the freely connected chain:

$$C_n = \frac{\langle R_n^2 \rangle}{n \cdot l^2} \quad \rightarrow \quad C_\infty = \frac{1 - \cos \varphi}{1 + \cos \varphi}$$

so, for a tetrahedral bond angle $\varphi = 109^\circ$: $C_\infty \approx 2$

Monte Carlo simulation of polymers

rotational isomeric state model (Flory, 1969)

- every link is always in one of defined rotational states
- these states, dihedral angles, are minima of pot. energy
- e.g. trans, gauche(+) and gauche(−) in polyalkane chain

Monte Carlo simulation of polymers

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- every link is always in one of defined rotational states
- these states, dihedral angles, are minima of pot. energy
- e.g. trans, gauche(+) and gauche(−) in polyalkane chain
- conformation-dependent properties described with matrices
- best known approximative description of polymer chains
- combined with MC → wide range of properties
 - conformations of chain are generated with probability distributions corresponding to their statistical weights, which are a component of the model (in a matrix form)

Monte Carlo simulation of polymers

rotational isomeric state model

matrix of statistical weights for an example of polyalkane chain:

$$U \equiv \begin{pmatrix} u_{tt} & u_{tg^+} & u_{tg^-} \\ u_{g^+t} & u_{g^+g^+} & u_{g^+g^-} \\ u_{g^-t} & u_{g^-g^+} & u_{g^-g^-} \end{pmatrix} = \begin{pmatrix} 1.00 & 0.54 & 0.54 \\ 1.00 & 0.54 & 0.05 \\ 1.00 & 0.05 & 0.54 \end{pmatrix}$$

u_{ab} – statistical weight of dihedral state b
following a link in the dihedral state a

Monte Carlo simulation of polymers

rotational isomeric state model

Starting on one end of the chain, a conformation is generated by calculating the dihedral angles sequentially, until the whole chain is done.

The probability of each dihedral angle is determined with **MC** using the a priori probabilities of the dihedral states and the state of the previous dihedral angle.

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In a typical study, a large number of such chain will be grown, and the properties of interest will be calculated and averaged. Estimated can be pair correlation functions, scattering functions and force–elongation profiles.